

Metabolically Engineered Micro-Organisms Having Reduced  
Production Of Undesired Metabolic Products

In the metabolic production of low-value added products like  
5 ethanol, lactic acid, citric acid, amino acids and many  
antibiotics, the yield of product on the substrate, i.e. the  
amount of product formed per unit substrate consumed (often  
given as kg product formed per kg substrate consumed), and  
the productivity, i.e. the amount of product formed per unit  
10 reaction volume and per unit time (often given as kg product  
formed per m<sup>3</sup> reaction volume per hour) are the most  
important design variables to optimise. In order to obtain a  
high yield and productivity it is necessary to direct the  
carbon fluxes from the substrate towards the metabolite of  
15 interest at a high rate and at the same time minimize the  
formation of all possible by-products. This often requires  
engineering of the central carbon metabolism, which is  
difficult due to the tight regulation in this part of the  
cellular metabolism (Nielsen, 2001).

20

Many fermentation processes have been optimised such that the  
desired product is predominantly formed. However, due to the  
complexity of cellular metabolism it is inevitable that by-  
products are formed. This may be undesirable for at least  
25 three reasons:

- The by-product(s) may be toxic to humans or animals.
- The by-product(s) may cause problems in the subsequent  
separation process or represent an environmental burden
- Formation of the by-product(s) results in a loss of  
30 carbon, and the overall yield of product on the raw  
material is therefore below the theoretical maximum.

The first reason is clearly problematic if humans or animals may be exposed to the product, either directly or indirectly. Typically, one chooses a cellular system that does not produce toxins when products are made for human consumption, e.g. fungal cells that do not produce aflatoxins are used for production of food grade enzymes, and *E. coli* strains that do not produce endotoxins are used for production of pharmaceuticals. The second reason may especially be problematic if the by-product has properties very similar to those of the desired product, since separation of the active product then requires very efficient separation principles, which are often costly. In some cases the by-product may also cause other types of problems, e.g. inactivation of the desired product. Problems with loss of carbon in the by-product is mainly a problem in connection with the production of low-value added products like ethanol, citric acid, amino acids and many antibiotics.

By way of example, a major problem in connection with ethanol production by anaerobic fermentation of *S. cerevisiae* is a substantial formation of glycerol as a by-product. Under anaerobic growth conditions cytosolic NADH formed from biomass formation can only be reconverted to  $\text{NAD}^+$  via glycerol formation (van Dijken and Scheffers, 1986). There are two genes, *GPD1* and *GPD2*, encoding glycerol-3-phosphate dehydrogenase that regenerates  $\text{NAD}^+$  from NADH while converting dihydroxyacetone-phosphate to glycerol-3-phosphate. Disruption of *GPD2* results in some reduction of glycerol formation, but the specific growth rate is also significantly reduced (Valadi et al., 1998; Nissen et al., 2000a). A double *gpd1gpd2* deletion mutant strain is not able to grow at anaerobic conditions, and introduction of a new

pathway to regenerate  $\text{NAD}^+$  was therefore attempted by expressing a bacterial transhydrogenase (catalysing:  $\text{NADH} + \text{NADP}^+ \rightleftharpoons \text{NAD}^+ + \text{NADPH}$ ) of *Azotobacter vinelandii* in a double *gpd1gpd2* deletion mutant strain (Nissen et al., 2000a).

- 5 Expression of the transhydrogenase could, however, not restore growth under anaerobic conditions. Enhanced expression of a transhydrogenase is disclosed also in EP-A-0733712.
- 10 WO99/46363 report on the expression of a phosphorylating dehydrogenase resulting in a net transhydrogenase activity in living cells with the aim to improve product formation. Thus, through over-expression of *GDH2*, encoding a phosphorylating NADH-dependent glutamate dehydrogenase (EC
- 15 1.2.1.12), they attempted to increase the ethanol formation from xylose and glucose. Another attempt using the same concept was to express a fungal NADPH dependent glyceraldehyde dehydrogenase as described by Verho et al. (2002). However, during growth on xylose the recombinant
- 20 cells also produced substantially more xylitol.

- In another approach to improve ethanol production and decrease glycerol formation Nissen et al. (2000b) engineered the ammonia assimilation in *S. cerevisiae*. Through
- 25 disruption of *GDH1*, encoding NADPH-dependent glutamate dehydrogenase, and over-expression of *GDH2* the production of NADH in association with biomass synthesis was reduced significantly resulting in a more than 40% reduction of the glycerol yield (Nissen et al., 2000b). Furthermore, through
- 30 over-expression of the GS-GOGAT pathway for ammonia assimilation, which is also NADH-dependent in *S. cerevisiae*, both a reduction in the glycerol yield and an increase in the

ethanol yield was obtained (Nissen et al., 2000b). This increase in ethanol yield is due to the additional consumption of ATP in the GS-GOGAT pathway. This example illustrates that redirection of the fluxes through the

5 central carbon metabolism can be obtained through engineering other parts of the metabolism, particularly through modulation of the redox metabolism, and this approach is likely to function also for improving the production of other products as the supply of NADPH for biosynthesis is often

10 limiting the capacity for production. While NADPH is almost exclusively used as electron donor in biosynthesis in the cell, NADH is primarily used for generation of free energy (often in the form of high-energy phosphate-bonds in ATP). Under some conditions, such as anaerobic growth of *S.*

15 *cerevisiae* on a fermentable sugar, surplus amounts of NADH are formed which cannot be used in generation of ATP, and this results in the formation of by-products, primarily glycerol.

20 Valverde et al (1999) discloses an *E. coli* strain engineered to express cDNA containing the *Pisum sativum* *GapN* gene which encodes the non-phosphorylating glyceraldehyde-3-phosphate dehydrogenase GAPN or GAPDHN (EC 1.2.1.9). The strain has its native *Gap-2* gene encoding its NAD-dependent

25 phosphorylating glyceraldehyde-3-phosphate dehydrogenase GAPDH disabled by an insertion. It is found that the expression of GAPN re-establishes the ability of the strain to grow aerobically on sugars, but the strain is still unable to perform anaerobic fermentation.

30

The present invention now provides a metabolically engineered micro-organism having an operative first metabolic pathway in

which a first metabolite is transformed into a second metabolite in a reaction in which NAD is a cofactor for a first enzyme, said reaction step producing NADH, and in which said second metabolite is transformed into at least one  
5 further metabolite in a reaction catalysed by a second enzyme, and having an operative second metabolic pathway characterised by an enzyme activity in excess of a native level in respect of a third enzyme catalysing a non-reversible reaction in which NADP is a cofactor and NADPH is  
10 a product and in which said first metabolite is transformed into a said further metabolite without the involvement of said second enzyme.

In a micro-organism of the invention as described above, said  
15 first metabolic pathway is preferably a native pathway.

In certain preferred embodiments, said first enzyme is a phosphorylating dehydrogenase.

20 In certain preferred embodiments, including those referred to immediately above, said second enzyme is a kinase.

In certain preferred embodiments, including those referred to in the two paragraphs immediately above, said third enzyme is  
25 a non-phosphorylating dehydrogenase, for instance said third enzyme is GAPN (EC 1.2.1.9).

An example of a phosphorylating dehydrogenase first enzyme is GAPDH (EC 1.2.1.12).

30

In certain preferred embodiments, including all those referred to above, at least one copy of a genetic sequence

encoding said third enzyme has been recombinantly introduced into said organism.

Preferably, a genetic sequence encoding said third enzyme is  
5 operatively linked to an expression signal not natively associated with said genetic sequence in said organism.

The micro-organism of the invention may preferably be a yeast. This may be an ethanol producing fermenting yeast.  
10 It may be a strain of *Saccharomyces cerevisiae*.

More generally, the micro-organism may be a species belonging to the genus *Saccharomyces*, e.g. *S. cerevisiae*, *S. kluyveri*, *S. bayanus*, *S. exiguus*, *S. sevazzi*, *S. uvarum*, a species  
15 belonging to the genus *Kluveromyces*, e.g. *K. lactis* *K. marxianus* var. *marxianus*, *K. thermotolerans*, a species belonging to the genus *Candida*, e.g. *C. utilis* *C. tropicalis*, a species belonging to the genus *Pichia*, e.g. *P. stipidis*, *P. pastoris*, *P. sorbitophila*, or other yeast  
20 species, e.g. *Debaromyces hansenii*, *Hansenula polymorpha*, *Yarrowia lipolytica*, *Zygosaccharomyces rouxii* or *Schizosaccharomyces pombe*.

Concerning other micro-organisms (non-yeast), a non-  
25 exhaustive list of suitable micro-organisms will include the following:

*Escherichia coli*, *Corynebacterium glutamicum*, *Aspergillus niger*, *Aspergillus awamori*, *Aspergillus oryzae*, *Aspergillus nidulans*, *Penicillium chrysogenum*, *Rhizopus oryzae*.

30

The invention includes a genetically transformed micro-organism containing one or more copies of an heterologous DNA

sequence encoding GAPN operatively associated with an expression signal and having a functional native or heterologous expression capability for GAPDH (EC 1.2.12).

5 In a further aspect, the invention includes a method of producing a desired metabolic product with decreased production of an undesired metabolic product, comprising culturing a micro-organism of the invention as described above. The undesired metabolic product may be glycerol,  
10 acetate or an amino acid, but it may also be any other metabolite secreted by micro-organisms.

The desired product may be ethanol, lactic acid, citric acid, an amino acid or an antibiotic.

15

The invention will therefore be useful for the improvement of production of ethanol but also of metabolites other than ethanol in micro-organisms. Thus, Porro et al. (1999) describe production of lactic acid in yeast through deletion  
20 of pyruvate decarboxylase activity and expression of a heterologous activity of lactate dehydrogenase. In the conversion of pyruvate to lactate there is regeneration of  $\text{NAD}^+$  as is the case in the overall conversion of pyruvate to ethanol. The overall conversion of a sugar to lactic acid  
25 therefore has high similarity with the conversion of a sugar to ethanol, and the invention will consequently have a positive effect on lactic acid production.

In the production of citric acid by the filamentous fungus  
30 *Aspergillus niger* there is a net formation of NADH in the conversion of sugar to citric acid, i.e. NADH is generated at the location of glyceraldehydes-3-P dehydrogenase and

pyruvate dehydrogenase. Through expression of GAPN it will be possible to replace part of the net formation of NADH with a net formation of NADPH, which is needed for protein synthesis. Hence, sugar that may otherwise be used to  
5 generate redox power in the form of NADPH, e.g. through the pentose phosphate pathway where sugar is converted to carbon dioxide with parallel formation of NADPH, may be redirected towards formation of citric acid resulting in a higher yield of the product on the sugar. Similar reasoning will hold for  
10 the production of other metabolites like succinic acid and malic acid.

In the production of many amino acids, e.g. lysine by *Corynebacterium glutamicum*, the amino acid is derived from  
15 precursor metabolites of the central carbon metabolism. Thus, in bacteria lysine is derived from oxaloacetate, which again is derived from pyruvate or phosphoenolpyruvate. In the conversion of sugar to the precursor metabolite there is a net production of NADH and the overall conversion of sugar  
20 to lysine therefore involves a net production of NADH. In the conversion of oxaloacetate to lysine there is a net consumption of NADPH (in some cases indirectly through the use of glutamate which needs to be regenerated from 2-oxoglutarate with expenditure of NADP). Expression of GAPN  
25 may therefore lead to a reduced net formation of NADH and a reduced net consumption of NADPH in the overall conversion of sugar to lysine. Similar reasoning will hold for the synthesis of other amino acids, e.g. isoleucine, threonine and phenylalanine.

30

In the production of many antibiotics, e.g. penicillin by *Penicillium chrysogenum*, there is also a net production of



NADH and a net consumption of NADPH in the overall conversion of sugar to antibiotics. For these processes expression of GAPN will therefore also be beneficial.

5 The enhanced expression of GAPN may be beneficial in combination with modulated (enhanced or suppressed) expression or activity of one or more other enzymes. Mention has been made above of interaction between GAPN expression and expression of lactate dehydrogenase in yeast. Generally,  
10 the invention may be used to improve the metabolism of pentose sugars. The invention may be used to improve xylose uptake and to reduce xylitol secretion. In many microorganisms xylose metabolism involves xylose reductase (XR), which converts xylose to xylitol, xylitol dehydrogenase  
15 (XDH), which converts xylitol to xylulose, and finally xylulose kinase that phosphorylates xylulose to xylulose-5-phosphate which enters the pentose phosphate pathway. XR involves formation of NADPH (the enzyme may use both  $\text{NAD}^+$  and  $\text{NADP}^+$  as co-factor, but it has preference for  $\text{NADP}^+$ ) whereas  
20 XDH involves consumption of NADH. There is therefore a net consumption of NADPH and a net formation of NADH upon xylose uptake. Thus, increased GAPN expression may, if it is linked with expression of xylose reductase, xylose dehydrogenase, and xylitol kinase, result in an increased  
25 xylose uptake.

Expression of GAPN or another said third enzyme may be provided for by the introduction into a micro-organism of one or more copies of a DNA coding sequence for the enzyme either  
30 with an heterologous promoter or placed under the control of an native promoter sequence. Suitably, the coding sequence

and an effective expression signal therefore is introduced in a multi-copy plasmid.

In preferred embodiments, this invention specifically targets  
5 the problem of production of surplus of NADH in functioning cells and the problem with a limited supply of NADPH. The cells are enabled to increase the formation of NADPH at the cost of NADH formation. This is done for example, by expression of a non-phosphorylating, NADP<sup>+</sup>-dependent  
10 glyceraldehyde-3-phosphate dehydrogenase (GAPN) (EC 1.2.1.9) in a cell, as a means to alter the redox metabolism in the cell.

GAPN catalyses the irreversible oxidation of glyceraldehyde-  
15 3-phosphate and NADP<sup>+</sup> into 3-phosphoglycerate and NADPH. In most cells the conversion of glyceraldehyde-3-phosphate into 3-phosphoglycerate is only catalysed by the sequential action of two enzymes, i.e. NAD<sup>+</sup>-dependent glyceraldehyde-3-phosphate dehydrogenase (GAPDH) (EC 1.2.1.12) and  
20 phosphoglycerate kinase (PGK) (EC 2.7.2.3) by conversion of NAD<sup>+</sup> and ADP into NADH and ATP (see figure 1). The net stoichiometry of the routes converting glyceralde-3-P into 3-P-glycerate are:

25           **GAPN:**            glyceraldehyde-3-phosphate + NADP<sup>+</sup> → 3-phosphoglycerate + NADPH

**GAPDH + PGK:** glyceraldehyde-3-phosphate + NAD<sup>+</sup> + ADP + P  
= 3-phosphoglycerate + NADH + ATP

30 The reaction catalysed by GAPN thereby yields one NADPH instead of one NADH and one ATP when comparing with the total reaction catalysed by GAPDH and PGK.

The conversion of glyceraldehyde-3-phosphate into 3-phosphoglycerate is part of the glycolysis, which is the main energy-yielding pathway, and the reaction is therefore always  
5 active when a cell is growing on hexose or pentose containing substrates.

By controlling the amount of *gapN* expressed in the microbial cell control can be exercised over how large a part of the  
10 conversion of glyceraldehyde-3-phosphate into 3-phosphoglycerate should go through GAPN. Thereby the redox metabolism can be manipulated by controlling the amount of one NADPH formed at the expense of one NADH and ATP.

15 The Examples below make reference to the accompanying drawings in which:

Figure 1 shows the results obtained in Example 8 in respect of the profiles of biomass, glucose and xylose concentrations (The symbols for xylose are filled circles and hollow  
20 diamonds and the two curves lie close overlapping one another); and

Figure 2 shows the results obtained in Example 8 in respect of the profiles of ethanol, xylitol, glycerol and acetate concentrations. (Closed and open square symbols are used for  
25 ethanol and for acetate. The lower closed and open square defined curves are acetate. The upper closed and open square defined curves are ethanol.)

To illustrate this in the Examples below, a nucleotide  
30 sequence containing the non-phosphorylating, NADP<sup>+</sup>-dependent glyceraldehyde-3-phosphate dehydrogenase gene (*gapN*) from *Streptococcus mutans* was expressed in *S. cerevisiae* on a

multicopy plasmid. The resulting strain was characterised in anaerobic batch cultivations on the hexose glucose - a typical method for ethanol production (Example 3) and on xylose (Example 8). NADP<sup>+</sup>-dependent glyceraldehyde-3-phosphate dehydrogenase activity was determined in Example 5 in both the *gapN* strain of Example 1 and in a reference strain carrying the empty plasmid. Activity of GAPN could only be measured in the *gapN* strain and the activity was approximately 10% of that of the NAD<sup>+</sup>-dependent glyceraldehyde-3-phosphate dehydrogenase activity.

The growth rate of the *gapN* strain of Example 1 was not affected by the expression of GAPN activity when comparing to the strain containing the empty plasmid. The *gapN* strain produced 43% less glycerol and 3% more ethanol. A much greater increase in ethanol production is achieved in Example 8.

Glycerol is formed by *S. cerevisiae* during anaerobic growth to maintain the cytosolic redox balance. Under anaerobic conditions NADH, produced as a result of production of biomass and organic acids, can only be oxidised to NAD<sup>+</sup> by formation of glycerol, since respiration is not possible and the formation of ethanol is a redox-neutral process. The formation of glycerol is therefore a redox problem, so by introducing *gapN* into *S. cerevisiae* the production of glycerol will be reduced by one molecule for each molecule of glyceraldehyde-3-phosphate that is converted via GAPN. By having a flux through GAPN that is high enough the production of glycerol can be completely eliminated and the flux redirected to ethanol and/or biomass thereby increasing the ethanol yield.

By increasing the amount of NADPH for biosynthesis by expressing *gapN* in a cell it can potentially be possible to increase the production of a product such as protein where  
5 the supply of NADPH might be limiting.

Advantages of the exemplified approach include:

- Larger changes in yields can be achieved.
- Only one genetic change is required.
- 10 • The activity of the enzyme only affects the specific reaction it catalysed and the redox metabolism - there are no effects on other parts of metabolism and therefore the growth rate is not affected.
- By-production of glycerol in production of ethanol with  
15 yeast can be eliminated.

**Example 1 Expression of GAPN in *S. cerevisiae*.**

**Strains:** *S. cerevisiae* (M4054, S288C MATa *ura3 gap1*) was used for construction of a reference strain and GAPN strain. For  
20 long-term maintenance plasmid bearing strains were grown to stationary phase in shake flask cultures on minimal media (see below). After addition of sterile glycerol to a concentration of 20% (vol/vol), aliquots were stored at -80°C. These frozen stocks were used for obtaining single  
25 colonies on plates with a minimal medium (Verduyn et al., 1990), which were stored at 4°C, and used within 2 weeks for inoculation of precultures.

**Construction of reference strain:** The empty pYX212 2 $\mu$  high-copy vector containing the *URA3* gene and the *TPI1* promoter was transformed into *S. cerevisiae* (M4054) by electroporation.

- 5 **Construction of GAPN strain:** *gapN* was expressed on a pYX212 2 $\mu$  high-copy vector containing the *URA3* gene and the *TPI1* promoter. The plasmid was constructed directly in *S. cerevisiae* (M4054) by cotransformation and homologous recombination between *EcoRI* digested pYX212 and PCR-amplified
- 10 *gapN* from *Streptococcus mutans*. PCR was performed on genomic DNA from *Streptococcus mutans* using Expand High Fidelity (Roche) and one primer identical to the *TPI1* promoter in pYX212 plus the first part of *gapN* (gapN-START-*EcoRI*-TPI promoter 5'-CTA CAA AAA ACA CAT ACA GGA ATT CAT GAC AAA ACA
- 15 ATA TAA AAA TTA TG) and a second primer (gapN-STOP-*NcoI*-BamHI-AvrII-ApaI 5'-GGG CCC TAG GAT CCA TGG TGA ATT TTA TTA TTT GAT ATC AAA TAC GAC GG) identical to the MCS of pYX212 and the last part of *gapN* including the stop codon. Hence the ORF of *gapN* has been cloned between *EcoRI* and *NcoI* side in
- 20 pYX212, down stream the *TPI1* promoter. The original start codon TTG was substituted with an ATG, to make translation in *S. cerevisiae* possible. Construction was verified by diagnostic PCR.

**Example 2 Shake flask cultivations and precultures:** Aerobic shake flask cultivations were performed in baffled, cotton-stopped, 500 ml Erlenmeyer flasks to screen transformants obtained. Precultures for anaerobic batch cultivations were grown in similar flasks but without baffles. These flasks contained each 100 ml of a defined mineral medium containing 7.5 g/L  $(\text{NH}_4)_2\text{SO}_4$ ; 14 g/L  $\text{KH}_2\text{PO}_4$ ; 0.5 g/L  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ ; 50  $\mu\text{L/L}$  antifoam (Sigma A-8436); 2% (w/vol) glucose; trace metals (15 mg/L EDTA; 4.5 mg/L  $\text{ZnSO}_4 \cdot 7\text{H}_2\text{O}$ ; 0.84 mg/L  $\text{MnCl}_2 \cdot 2\text{H}_2\text{O}$ ; 0.30 mg/L  $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$ ; 0.30 mg/L  $\text{CuSO}_4 \cdot \text{H}_2\text{O}$ ; 0.40 mg/L  $\text{Na}_2\text{MoO}_4 \cdot 2\text{H}_2\text{O}$ ; 4.5 mg/L  $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ ; 3.0 mg/L  $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$ ; 1.0 mg/L  $\text{H}_3\text{BO}_3$ ; and 0.10 mg/L KI) and vitamins (0.05 mg/L D(-)biotin; 1.0 mg/L Ca D(+)panthotenate; 1.0 mg/L nicotinic acid; 25 mg/L myo-inositol; 1.0 mg/L thiamine chloride hydrochloride; 1.0 mg/L pyridoxol hydrochloride; and 0.20 mg/L p-aminobenzoic acid). The pH of the mineral medium was set to 6.5 with NaOH and autoclaved separately from the glucose solution. After autoclavation the vitamin solution was added to the flasks by sterile filtration. Shake flasks and precultures were inoculated with a single colony from plate cultures grown at 30°C and 150 rpm. Precultures were grown to exponential phase and used for inoculation of anaerobic batch cultivations to a start concentration of 1 mg CDW/L.

Growth of both the reference and the GAPN-strain was observed.

**Example 3 Anaerobic batch cultivations:** Cultivations were

5 carried out in well-controlled laboratory fermentors (B. Braun Biotech, Germany) with a working volume of 2 litres. A defined mineral medium (Verduyn et al., 1990) was used, which contained per litre: 40 g glucose; 5.0 g  $(\text{NH}_4)_2\text{SO}_4$ ; 3.0 g  $\text{KH}_2\text{PO}_4$ ; 0.5 g  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ ; and trace metals and vitamins as  
10 described in shake flask cultivations and precultures. 300  $\mu\text{l/L}$  antifoam (Sigma A-8436) was added to avoid foaming and the medium was supplemented with 420 mg/L Tween 80 and 10 mg/L ergosterol, which is necessary for anaerobic growth of *S. cerevisiae*. The glucose solution was autoclaved separately  
15 from the mineral medium and afterwards added to the fermentor together with a sterile filtrated solution containing the vitamins and together with the Tween 80 and ergosterol, which first were dissolved in boiling pure ethanol.

Cultivations were carried out at 30°C with a stirrer speed of  
20 600 rpm and were flushed with nitrogen gas at a flow rate of 400 ml per minute. To minimise the diffusion of  $\text{O}_2$  into the cultures, the bioreactors were fitted with Norprene tubing. The concentration of dissolved oxygen was measured with Mettler Toledo polarographic electrode and remained below the



detection limit. pH was kept at 5.0 by automatic addition of 4 M KOH. The bioreactors were fitted with cooled condensers, and the off-gas was led to a gas analyser (INNOVA, Denmark) to measure the content of CO<sub>2</sub>. The biomass concentration during fermentation with the reference and GAPN strains are shown in Figure 2. It is found that the maximum specific growth rate of the GAPN strain is identical with that of the reference strain.

**Example 4 Analysis of extracellular metabolites:** Culture samples for determination of glucose, ethanol, glycerol, acetate, pyruvate and succinate concentrations were filtered through a 0.45 µm cellulose acetate filter (Osmonics) immediately after sampling, and the filtrate was frozen at -20°C until further analysis. The concentrations of the metabolites were determined by high-pressure liquid chromatography on an Aminex HPX-87Hm column (Bio-Rad) kept at 65°C and eluted at 0.6 ml per minute with 5 mM H<sub>2</sub>SO<sub>4</sub>. Acetate and pyruvate were detected spectrophotometrically by a Waters 486 Turnable Absorbance Detector at 210 nm. Glucose, ethanol, glycerol and succinate were detected refractometrically by a Waters 410 Differential Refractometer. Measurement of the metabolites during anaerobic fermentations is shown in Figure

3. The final concentrations of the metabolites are listed below. It is seen that the GAPN strain produces more ethanol and less glycerol.

5

	Ethanol (g/L)	Glycerol (g/L)	Pyruvate (g/L)
Reference strain	14,90	2,81	0,061
GAPN strain	17,05	2,06	0,066

Based on the measurements of glucose the overall conversion yields to the different metabolites were calculated and they are given below.

10

	Ethanol (g/g glc)	Glycerol (g/g glc)	Pyruvate (g/g glc)
Reference strain	0,392	0,078	0,0013
GAPN strain	0,403	0,0463	0,0012

It is observed that the ethanol yield is improved 3% and the glycerol yield is reduced by 40%.

15

**Example 5 Measurement of enzyme activities:** Cell free extracts were produced with the help of a Fastprep FP120 instrument (Savant Instruments, New York) as described by Møller *et al.* (2002).

5 Enzyme activities were assayed at 30°C by following the NADPH or NADH production at 340 nm using a spectrophotometer (HP 8353 UV-VIS system with Chemstation software from Hewlett Packard). Glyceraldehyde-3-phosphate dehydrogenase activity was determined as described by Crow and Wittenberger (1979)  
10 in a 1 ml reaction mixture activity. For determination of non-phosphorylating, NADP<sup>+</sup>-dependent glyceraldehyde-3-phosphate dehydrogenase activity the reaction mixture contained: 125 mM triethanolamine/HCl buffer (pH 8.3), 1 mM NADP<sup>+</sup>, 5 mM 2-mercaptoethanol, and cell free extract. NAD<sup>+</sup>-  
15 dependent glyceraldehyde-3-phosphate dehydrogenase activity was determined with a reaction mixture containing: 125 mM triethanolamine/HCl buffer (pH 8.3), 1 mM NAD<sup>+</sup>, 5 mM cystein/HCl, and cell free extract. The reactions were started by adding DL-glyceraldehyde-3-phosphate (prepared  
20 from DL-glyceraldehyde-3-phosphate diethyl acetal, Sigma G-5376) to a final concentration of 2 mM. Protein content in cell free extracts was determined by the Lowry method, using fatty-acid free BSA (Sigma A-6003) as standard. Results from analysis of the enzyme activity are given below. It is seen

that in the reference strain there is no activity of a NADP-dependent glyceraldehydes dehydrogenase whereas in the GAPN strain there is some low activity (accounting for about 10% of the total glyceraldehydes dehydrogenase activity).

5

	Reference strain	GAPN strain
NAD-dependent glyceraldehyde dehydrogenase	11,2 ± 0,74	7,03 ± 0,71
NADP-dependent glyceraldehyde dehydrogenase	0	0,65 ± 0,03

#### Example 6 Sequencing of *GapN*

10 In order to verify that no mutations had occurred in the *gapN* encoding gene expressed in the 2 $\mu$  high copy number plasmid, the part of the plasmid containing the *gapN* was sequenced. The sequence of the *gapN* as inserted into the plasmid and as found actually to be present therein was 'atg aca aaa caa tat

15 aaa aat tat gtc aat ggc gag tgg aag ctt tca gaa aat gaa att  
aaa atc tac gaa ccg gcc agt gga gct gaa ttg ggt tca gtt cca

gca atg agt act gaa gaa gta gat tat gtt tat gct tca gcc aag  
aaa gct caa cca gct tgg cga tca ctt tca tac ata gaa cgt gct  
gcc tac ctt cac aag gta gca gat att ttg atg cgt gat aaa gaa  
aaa ata ggt gct gtt ctt tcc aaa gag gtt gct aaa ggt tat aaa  
5 tca gca gtc agc gaa gtt gtt cgt act gca gaa atc att aat tat  
gca gct gaa gaa ggt ctt cgt atg gaa ggt gaa gtc ctt gaa ggc  
ggc agt ttt gaa gca gcc agc aag aaa aaa att gcc gtt gtt cgt  
cgt gaa cca gta ggt ctt gta tta gct att tca cca ttt aac tac  
cct gtt aac ttg gca ggt tcg aaa att gca ccg gct ctt att gcg  
10 gga aat gtt att gct ttt aaa cca ccg acg caa gga tca atc tca  
ggg ctc tta ctt gct gaa gca ttt gct gaa gct gga ctt cct gca  
ggg gtc ttt aat acc att aca ggt cgt ggt tct gaa att gga gac  
tat att gta gaa cat caa gcc gtt aac ttt atc aat ttt act ggt  
tca aca gga att ggg gaa cgt att ggc aaa atg gct ggt atg cgt  
15 ccg att atg ctt gaa ctc ggt gga aaa gat tca gcc atc gtt ctt  
gaa gat gca gac ctt gaa ttg act gct aaa aat att att gca ggt  
gct ttt ggt tat tca ggt caa cgc tgt aca gca gtt aaa cgt gtt  
ctt gtg atg gaa agt gtt gct gat gaa ctg gtc gaa aaa atc cgt  
gaa aaa gtt ctt gca tta aca att ggt aat cca gaa gac gat gca  
20 gat att aca ccg ttg att gat aca aaa tca gct gat tat gta gaa  
ggg ctt att aat gat gcc aat gat aaa gga gcc act gcc ctt act  
gaa atc aaa cgt gaa ggt aat ctt atc tgt cca atc ctc ttt gat  
aag gta acg aca gat atg cgt ctt gct tgg gaa gaa cca ttt ggt  
cct gtt ctt ccg atc att cgt gtg aca tct gta gaa gaa gcc att

gaa att tct aac aaa tcg gaa tat gga ctt cag gct tct atc ttt  
aca aat gat ttc cca cgc gct ttt ggt att gct gag cag ctt gaa  
gtt ggt aca gtt cat atc aat aat aag aca cag cgc ggc acg gac  
aac ttc cca ttc tta ggg gct aaa aaa tca ggt gca ggt att caa  
5 ggg gta aaa tat tct att gaa gct atg aca act gtt aaa tcc gtc  
gta ttt gat atc aaa', which is identical with the sequence of  
*gapN* from *Streptococcus mutans*.

**Example 7 Expression of GAPN in a xylose metabolising strain  
10 of *S. cerevisiae***

**Strains:** *S. cerevisiae* (MATa *SUC2 MAL2-8 pADH-XYL1 pPGK-XYL2*  
*pPGK-XKS1 ura3*<sup>-</sup>) was used for construction of a xylose  
metabolising strain of *S. cerevisiae* expressing GAPN. For  
long-term maintenance, plasmid bearing strains were grown to  
15 stationary phase in shake flask cultures on minimal media  
(see below). After addition of sterile glycerol to a  
concentration of 20% (vol/vol), aliquots were stored at -  
80°C. These frozen stocks were used for obtaining single  
colonies on plates with a minimal medium (Verduyn et al.,  
20 1990), which were stored at 4°C, and used within 2 weeks for  
inoculation of precultures.

**Construction of reference strain:** The empty pYX212 2μ high-  
copy vector containing the *URA3* gene and the *TPI1* promoter

was transformed into *S. cerevisiae* (*MATa SUC2 MAL2-8 pADH-XYL1 pPGK-XYL2 pPGK-XKS1 ura3<sup>-</sup>*) by the LiAc method and positive clones, containing the pYX212 *URA3* 2 $\mu$  plasmid, were selected on synthetic minimal ammonium medium with glucose.

- 5    **Construction of GAPN strain:** The pYX212 2 $\mu$  high-copy vector containing the *URA3* gene and the *TPI1* promoter in front of the *gapN* encoding gene was transformed into *S. cerevisiae* (*MATa SUC2 MAL2-8 pADH-XYL1 pPGK-XYL2 pPGK-XKS1 ura3<sup>-</sup>*) by the LiAc method, and positive clones, containing *gapN* down stream  
10 the *TPI1* promoter in the *URA3* 2 $\mu$  plasmid, were selected on synthetic minimal ammonium medium with glucose.

A number of transformants was tested in aerobic shake flasks on glucose for growth rate and GAPN activity. A transformant was chosen with unchanged growth rate and high GAPN activity  
15 and denoted Xyl GAPN strain - the reference strain was denoted Xyl reference strain. Both strains had similar xylose reductase and xylose dehydrogenase activities, which were within the expected range.

- 20    **Example 8 Anaerobic batch cultivations of a xylose metabolising strain of *Saccharomyces cerevisiae*:** Cultivations were carried out in well-controlled laboratory fermentors with a working volume of 4 litres. A defined mineral medium

(Verduyn et al., 1990) was used, which contained per litre:  
20 g glucose; 50 g xylose; 5.0 g  $(\text{NH}_4)_2\text{SO}_4$ ; 3.0 g  $\text{KH}_2\text{PO}_4$ ; 0.5  
g  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ ; and trace metals and vitamins as described in  
shake flask cultivations and precultures. 500  $\mu\text{l/L}$  antifoam  
5 (Sigma A-8436) was added to avoid foaming and the medium was  
supplemented with 420 mg/L Tween 80 and 10 mg/L ergosterol,  
which is necessary for anaerobic growth of *S. cerevisiae*. The  
glucose solution was autoclaved separately from the mineral  
medium and afterwards added to the fermentor together with a  
10 sterile filtered solution containing the vitamins and  
together with the Tween 80 and ergosterol, which first were  
dissolved in boiling pure ethanol.

Cultivations were carried out at 30°C with a stirrer speed of  
500 rpm and were flushed with nitrogen gas at a flow rate of  
15 800 ml per minute. To minimise the diffusion of  $\text{O}_2$  into the  
cultures, the bioreactors were fitted with Norprene tubing.  
pH was kept at 5.0 by automatic addition of 2 M NaOH. The  
bioreactors were fitted with cooled condensers, and the off-  
gas was led to a gas analyser (INNOVA, Denmark) to measure  
20 the content of  $\text{CO}_2$ . The profiles of the concentrations of  
biomass, glucose and xylose for both the Xyl reference strain  
(closed symbols) and the Xyl GAPN strain (open symbols) are  
shown in Figure 1. The profiles of the concentrations of  
ethanol, xylitol, glycerol and acetate for both the Xyl



reference strain (closed symbols) and the Xyl GAPN strain (open symbols) are shown in Figure 2. It is found that the glucose and xylose uptake is approximately the same in the two strains whereas the Xyl GAPN strain produces much more ethanol, much less xylitol and less glycerol compared with the Xyl reference strain. The Xyl GAPN strain produces slightly more acetate than the Xyl reference strain. After 50 hours of fermentation the following concentrations of the metabolic products were obtained:

10

	<b>Ethanol</b> (g/L)	<b>Glycerol</b> (g/L)	<b>Xylitol</b> (g/L)
Xyl reference strain	12.3	3.5	13.4
Xyl GAPN strain	14.1	1.9	7.2

This corresponds to a 15% increase in the ethanol concentration, a 45% decrease in glycerol concentration and a 45% decrease in xylitol concentration.

15

During growth on glucose there is a slightly higher production of ethanol and a smaller production of glycerol by the Xyl GAPN strain (in analogy with results of Example 4). During growth on xylose overall the yield coefficients were estimated and are given below.

20

	Ethanol (g/g sugar)	Glycerol (g/g sugar)	Xylitol (g/g sugar)
Xyl reference strain	0.22	0.065	0.33
Xyl GAPN strain	0.29	0.037	0.17

These yield coefficients corresponds approximately with a 30% increase in the ethanol yield, a 40% decrease in glycerol  
5 yield and an almost 50% decrease in xylitol yield.

**Example 9 Anaerobic chemostat cultivations of a xylose  
metabolising strain of *Saccharomyces cerevisiae***

10 Anaerobic steady-state chemostat cultures were obtained in well-controlled 2-litre jacketed bioreactors (B. Braun Biotech, Melsungen, Germany) with a constant working volume of 1.0 litre and a dilution rate of  $0.05 \text{ h}^{-1}$  (which equals the specific growth rate). Cultivations were carried out at  
15  $30^{\circ}\text{C}$  with a stirrer speed of 500 rpm and were flushed with pure nitrogen gas ( $0.2 \text{ liter}\cdot\text{min}^{-1}$ ). pH was kept at 5.0 by automatic addition of 2 M NaOH. To minimise the diffusion of  $\text{O}_2$  into the cultures, the bioreactors were fitted with Norprene tubing. The concentration of dissolved oxygen was  
20 measured with Mettler Toledo polarographic electrode and

remained below the detection limit. The bioreactors were fitted with cooled condensers (5°C), and the off-gas was led to a gas analyser (INNOVA, Denmark) to measure the content of CO<sub>2</sub>. Steady state was assumed when at least five residence  
5 times had passed since starting the continuous cultivation and was verified by constant ( $\pm 2\%$ ) biomass concentration, metabolite concentrations, and carbon dioxide evolution rate at two measurements taken over 2 residence times. The cultures were fed with a defined medium similar to that  
10 described in Example 8.

Precultures for chemostat cultivations were grown at 30°C and 150 rpm for 16-24 hours in cotton-stopped, 500 ml Erlenmeyer flasks with baffles containing each 100 ml of a media of pH 6.5 similar to that in the fermenters, except from different  
15 concentrations of glucose (20 g.l<sup>-1</sup>), (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> (7.5 g.l<sup>-1</sup>), KH<sub>2</sub>PO<sub>4</sub> (14 g.l<sup>-1</sup>), and antifoam (50 µl.l<sup>-1</sup>), and furthermore, no Tween or ergosterol were added. Precultures were inoculated from plate cultures grown at 30°C. Continuous cultures were inoculated with 20 ml exponential growing preculture.

20

In the chemostat culture at steady state the concentrations of ethanol, glycerol and xylitol were measured, and the results are given below.

	<b>Ethanol</b> <b>(g/L)</b>	<b>Glycerol</b> <b>(g/L)</b>	<b>Xylitol</b> <b>(g/L)</b>
Xyl reference strain	10.84	1.49	7.41
Xyl GAPN strain	12.05	0.55	4.53

Based on the measurements of sugars the overall conversion yields to the different metabolites were calculated and they  
5 are given below.

	<b>Ethanol</b> <b>(g/g sugar)</b>	<b>Glycerol</b> <b>(g/g sugar)</b>	<b>Xylitol</b> <b>(g/g sugar)</b>
Xyl reference strain	0.29	0.044	0.21
Xyl GAPN strain	0.36	0.018	0.14

It is observed that the ethanol yield is improved by 24% and the glycerol yield is reduced by almost 60% and that the  
10 xylitol yield is reduced by 33%.

In this specification, unless expressly otherwise indicated, the word 'or' is used in the sense of an operator that returns a true value when either or both of the stated conditions is met, as opposed to the operator 'exclusive or'  
15 which requires that only one of the conditions is met. The

word 'comprising' is used in the sense of 'including' rather than in to mean 'consisting of'.

## References:

- Nielsen, J. (2001). Metabolic engineering. Appl. Microbiol. Biotechnol. 55, 263-283.
- 5 Nissen, T.L., Hamann, C.W., Kielland-Brandt, M.C., Nielsen, J., and Villadsen, J. (2000a). Anaerobic and aerobic batch cultivations of *Saccharomyces cerevisiae* mutants impaired in glycerol synthesis. Yeast 16, 463-474.
- Nissen, T.L., Kielland-Brandt, M.C., Nielsen, J., and  
10 Villadsen, J. (2000b). Optimization of ethanol production in *Saccharomyces cerevisiae* by metabolic engineering of the ammonium assimilation. Metabol. Eng. 2, 69-77.
- Porro, D., Bianchi, M.M., Brambilla, L., Menghini, R., Bolzani, D., Carrera, V., Lievenze, J., Liu, C.L., Ranzi, B.M.,  
15 Frontali, L., and Alberghina, L. (1999). Replacement of a metabolic pathway for large-scale production of lactic acid from engineered yeasts. Appl. Environ. Biotechnol. 65, 4211-4215.
- Valadi, H., Larsson, C., and Gustafsson, L. (1998). Improved  
20 ethanol production by glycerol-3-phosphate dehydrogenase mutants of *Saccharomyces cerevisiae*. Appl. Microbiol. Biotechnol. 50, 434-439.
- Valverde, F., Losada, M., and Serrano, A. (1999). Engineering a  
25 central metabolic pathway: glycolysis with no net phosphorylation in an *Escherichia coli* gap mutant complemented with a plant GapN gene. FEBS Lett. 449, 153-158.

Van Dijken, J.P. and Scheffers, W.A. (1986). Redox balances in the metabolism of sugars by yeasts. FEMS Microbiol. Rev. 32 , 199-224.

Verduyn, C., Postma, E., Scheffers, W.A., and van Dijken, J.P.  
5 (1990). Physiology of *Saccharomyces cerevisiae* in anaerobic glucose- limited chemostat cultures. J. Gen. Microbiol. 136, 395-403.

Verho, R., Richard, P., Jonson, P. H., Sundqvist, L.,  
Londesborough, J., and Penttilä, M. (2002) Identification of  
10 the first fungal NADP-GAPDH from *Kluyveromyces lactis*.  
Biochem. 41, 13833-13838